Paper / Subject Code: 49101 / ENGINEERING CHEMISTRY-I

1T00513 - S.E.(CHEMICAL)(Sem III) (CBSGS) / 49101 - ENGINEERING CHEMISTRY-I

Duration: 3 Hours Mark		arks: 80
N.B. : (1)	Question No.1 is compulsory.	43999
(2)	Attempt any three questions from remaining five questions.	
Q.1. Ansv	wer any fou r of the following.	20
A) D	raw and explain structure of Fe(CO) ₅ .	9000 00 00 00 00 00 00 00 00 00 00 00 00
,	rite IUPAC names of the following compounds	
	$[Zn(OH)_4]^{2-}$	
ii)	$K_3[Al(CO_4)_3]$	
C) W	rite the mechanism of Michael reaction.	W. D. C. C. C.
D) D	istinguish between SN ¹ and SN ² reactions.	
E) E:	xplain kinetically controlled reactions w.r.t. nitration of chlorobenzene.	
F)	Write a note on hydrogen bonding.	B. F. B. B.
Q.2.A) W	rite the Chemical formulae of the following coordination compounds	05
(i)	Hexacyanocobaltate (III) ion.	
(ii)	Pentaaquahydroxoaluminium (III) ion.	
	plain electrophilic substitution reaction w.r.t. Friedel Craft alkylation.	05
	aw molecular orbital diagram of nitrogen molecule. Calculate its bond order	
	mment on its magnetic properties.	05
D) Di	stinguish between transition state and intermediates.	05
Q.3.A) W	hat is EAN? Calculate EAN of [PtCl ₆] ²⁻	05
	plain the role of Zn in biological reactions.	05
	fferentiate between bonding and antibonding molecular orbitals.	05
D) Ex	plain stability of carbon free radicals w.r.t. inductive effect and resonance	effect. 05
Q.4.A) Ex	aplain the structure of SF ₄ on the basis of VSEPR theory.	05
	plain chemistry of cytochromes and their applications.	05
	hat is CSFE? Calculate CFSE of d ⁴ and d ⁷ in octahedral complexes.	05
D) W1	rite short note on elimination reactions.	05
Q.5.A) W	rite mechanism of Wohl – Ziegler allylic bromination.	05
B) Dr	aw M.O. diagram of CO molecule & calculate its bond order.	05
C) Ex	plain oxygen atom transfer biomolecular reaction containing iron.	05
D) W1	rite limitations of CFT.	05
Q.6.A) Ex	aplain with mechanism electrophilic substitution in monocyclic aromatic co	ompound. 05
B) Wl	hat is geometrical isomerism? Explain it in coordination compounds w.r.t. mber 6.	-
	rite the drawbacks of VBT.	05
	plain the structure of singlet carbenes. Comment on the stability of carbene	