University of Mumbai Online Examination 2020

Program: BE Chemical Engineering Curriculum Scheme: Revised 2016 Examination: Third Year Semester VI Course Code: CHC602 Course Name: Mass Transfer Operation II

Q1.	Which of the following is not a step in the process of distillation?	
Option A:	vaporization	
Option B:	condensation	
- F		
Option C:	heating	
Option D:	precipitation	
Ans:		
Alls.		
Q2.	In the distillation column, the temperature is highest at	
Q2.	In the distinution column, the temperature is ingliest at	
Option A:	top of the column	
Option B:	bottom of the column	
Option C:	In the middle of column	
Option D:	near wall of the entire column	
Option D.		
Ans:		
Q3.	Separation of two volatile liquids by distillation makes use of their	
Option A:	selectivity	
Option B:	density difference	
Option C:	relative volatility	
Option D:	solubility	
Ans:		
Q4.	Minimum reflux ratio in a distillation column results in	
Option A:	Maximum condenser size	
Option B:	Minimum reboiler size	
Option C:	Optimum number of trays	
Option D:	Minimum number of trays	
Ans:		
05		
Q5.	In Azeotropic mixture, the equilibrium vapour composition is	
Option A:	More than liquid composition	
Option B:	Independent of pressure	
Option C:	Same as liquid composition	
Option D:	Independent of pressure	
Ans:		
06	Longon value of the distribution coefficient	
Q6.	Larger value of the distribution coefficient	

Option A:	More is the solvent required		
Option B:	Less is the solvent required		
Option C:	There is no effect of the amount of solvent used		
Option D:	Is not desired		
Ans:			
Q7.	In the first stage of a three stage cross current liquid extraction, 100 kg of feed solution containing 15 % weight of solute was contacted with 75 kg of fresh solvent. if the solvent is completely immiscible with the feed solution, the slope of the operating line is		
Option A:	1.13		
Option B:	-1.13		
Option C:	2.3		
Option D:	-2.3		
Ans:			
Q8.	Common Leaching solvent can be		
Option A:	water		
Option B:	Sulphuric acid		
Option C:	Hydrogen chloride		
Option D:	Sodium Hydroxide		
Ans:	· ·		
Q9.	Tea percolation employs		
Option A:	Distillation		
Option B:	Absorption		
Option C:	Leaching		
Option D:	Drying		
Ans:			
Q10.	For better Leaching, the viscosity of the solvent should be		
Option A:	Less		
Option B:	High		
Option C:	No effect		
Option D:	Very high		
Ans:			
Q11.	Sugar recovery from sugar beats is by		
Option A:	Absorption		
Option B:	Leaching		
Option C:	Distillation		
Option D:	Adsorption		
Ans:			
Q12.	Which one of the following characteristics is not correct for physical adsorption		
Option A:	Adsorption on solids is reversible		
Option B:	Adsorption increases with increase in temperature		
Option C:	Adsorption is spontaneous		
Option D:	Both enthalpy and entropy of adsorption are negative		
Ans:			

Q13.	Which of the following not true for Freundlich isotherm		
Option A:	The isotherm is applicable in certain limit of pressure		
Option B:	constants k and n changes with temperature		
Option D:	It shows deviation at low pressure		
Option D:	Freundlich isotherm is empirical		
Ans:			
Alls.			
Q14.	Breakpoint time		
Option A:	increases with decrease in bed height		
Option B:	decreases with decreases bed height		
Option D: Option C:	not affected by bed height		
Option D:	first increases and then decreases with bed height		
Ans:			
Alls.			
Q15.	with increases in flow rate		
Option A:	break point time decreases		
Option A: Option B:	breakpoint time increases		
Option C:	breakpoint time decreases first and then increases		
Option D:	breakpoint is not affected		
Ans:			
Ans:			
016	UTU for peaked had adapther is		
Q16.	HTU for packed bed adsorber is Heat of transfer units		
Option A: Option B:	Heat of Unit		
Option B: Option C:			
-	Height of temperature unit		
Option D: Ans:	Height of transfer unit		
Alls.			
Q17.	One of the most common solvent used in crysatllisation are		
Option A:	Water		
Option B:	Alcohol		
Option C:	Normal saline		
Option D:	Sulphuric acid		
Ans:			
1 1115.			
Q18.	Crystalline solids can be recognised by their		
Option A:	low boiling point		
Option B:	sharp melting point		
Option C:	color		
Option D:	moderate melting point		
Ans:			
Q19.	The insoluble impurities from solution during crystallisation are removed by		
Option A:	Drying		
Option B:	Filtration		
Option C:	Heating		
Option D:	Cooling		
Ans:			
Q20.	What is the molar transmembrane flux of species if Premeability $P=10$		
	1		

	units,Membrane thickness l= 10units,Driving force = 200units		
Option A:	500		
Option B:	200		
Option C:	400		
Option D:	600		
Ans:			
Q21.	Which of the following is not an application of transport in membranes?		
Option A:	Microfiltration		
Option B:	Reverse osmosis		
Option C:	Dialysis		
Option D:	Fractional distillation		
Ans:			
Q22.	If the pressure drop(ΔP) is 1000 units, the flux(J) is 50 units, what is the hydraulic		
	membrane permeability?		
Option A:	0.02		
Option B:	0.04		
Option C:	0.05		
Option D:	0.06		
Ans:			
Q23.	Which of the following is not true about membrane separations?		
Option A:	Components which are passed through the membrane is called permeate		
Option B:	Components which are not passed through are called retentate		
Option C:	Non-porous membrane is never used		
Option D:	Membrane separations require a driving force		
Ans:			
Q24.	What is the reflux ratio at total reflux		
Option A:	0		
Option B:	∞		
Option C:	1		
Option D:	0.5		
Ans:			
Q25.	is a result of chemical interaction between solid and the adsorbed		
	substance		
Option A:	elution		
Option B:	desorption		
Option C:	chemisorption		
Option D:	physical adsorption		
Ans:			

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Question	Correct Option
	(Enter either 'A' or 'B' or 'C' or 'D'
Q1.	D
Q2.	В
Q3.	С
Q4	A
Q5	С
Q6	В
Q7	В
Q8.	Α
Q9.	С
Q10.	Α
Q11.	В
Q12.	В
Q13.	С
Q14.	В
Q15.	С
Q16.	D
Q17.	Α
Q18.	В
Q19.	В
Q20.	В
Q21.	D
Q22.	C
Q23.	C.
Q24.	В
Q25.	C